

09/10/91  
WO 95/21151

Rec'd PCT/PTO 29 SEP 1995

~~08/552777~~  
PCT/EP95/00221  
08/926835

- 1 -

Process for the hydrogenation of imines

The present invention relates to a process for the hydrogenation of imines with hydrogen under elevated pressure in the presence of iridium catalysts and a halide, wherein the reaction mixture contains an inorganic or organic acid.

US-A-4 994 615 describes a process for the asymmetric hydrogenation of prochiral N-arylketimines wherein iridium catalysts having chiral diphosphine ligands are used. US-A-5 011 995 describes a process for the asymmetric hydrogenation of prochiral N-alkylketimines using the same catalysts. US-A-5 112 999 discloses polynuclear iridium compounds and a complex salt of iridium, which contain diphosphine ligands, as catalysts for the hydrogenation of imines.

Those homogeneous catalysis processes have proved valuable, although it is evident, especially in the case of relatively large batches or on an industrial scale, that the catalysts frequently tend to become deactivated to a greater or lesser extent depending on the catalyst precursor, the substrate and the diphosphine ligands that are used. In many cases, especially at elevated temperatures - for example at temperatures  $>25^{\circ}\text{C}$ , which are necessary for a short reaction time - it is not possible to achieve complete conversion. For industrial applications of the hydrogenation process, therefore, the catalyst productivity is too low from the point of view of economic viability.

It has now been found, surprisingly, that the catalyst activity can be increased by a factor of 10 or more if the reaction mixture essentially contains a halide and also contains an acid. It has also unexpectedly been found that at the same time the deactivation of the catalysts can be considerably reduced or completely eliminated. It has also been found, surprisingly, that the enantioselectivity under the chosen conditions is high, and high optical yields of, for example, up to 80 % can be achieved, even at reaction temperatures of more than  $50^{\circ}\text{C}$ .

The invention relates to a process for the hydrogenation of imines with hydrogen under elevated pressure in the presence of iridium catalysts and with or without an inert solvent, wherein the reaction mixture contains an ammonium chloride, bromide or iodide, or a metal chloride, bromide or iodide that is soluble in the reaction mixture, the metal preferably being an alkali metal, and additionally contains an acid.

- 2 -

Suitable imines are especially those that contain at least one  $\text{>C=N—}$  group. If the groups are substituted asymmetrically and are thus compounds having a prochiral ketimine group, it is possible in the process according to the invention for mixtures of optical isomers or pure optical isomers to be formed if enantioselective or diastereoselective iridium catalysts are used. The imines may contain further chiral carbon atoms. The free bonds in the above formulae may be saturated with hydrogen or organic radicals having from 1 to 22 carbon atoms or organic hetero radicals having from 1 to 20 carbon atoms and at least one hetero atom from the group O, S, N and P. The nitrogen atom of the group  $\text{>C=N—}$  may also be saturated with  $\text{NH}_2$  or a primary amino group having from 1 to 22 carbon atoms or a secondary amino group having from 2 to 40 carbon atoms. The organic radicals may be substituted, for example, by F, Cl, Br,  $\text{C}_1\text{--C}_4$ haloalkyl wherein halogen is preferably F or Cl,  $\text{—CN}$ ,  $\text{—NO}_2$ ,  $\text{—CO}_2\text{H}$ ,  $\text{—CONH}_2$ ,  $\text{—SO}_3\text{H}$ ,  $\text{—PO}_3\text{H}_2$ , or  $\text{C}_1\text{--C}_{12}$ alkyl esters or amides, or by phenyl esters or benzyl esters of the groups  $\text{—CO}_2\text{H}$ ,  $\text{—SO}_3\text{H}$  and  $\text{—PO}_3\text{H}_2$ . Aldimine and ketimine groups are especially reactive, with the result that using the process according to the invention it is possible selectively to hydrogenate  $\text{>C=N—}$  groups in addition to the  $\text{>C=C<}$  and/or  $\text{>C=O}$  groups. Aldimine and ketimine groups are also to be understood to include  $\text{>C=N—N—}$  hydrazone groups.

The process according to the invention is suitable especially for the hydrogenation of aldimines, ketimines and hydrazones with the formation of corresponding amines and hydrazines, respectively. The ketimines are preferably N-substituted. It is preferable to use chiral iridium catalysts and to hydrogenate enantiomerically pure, chiral or prochiral ketimines to prepare optical isomers, the optical yields (enantiomeric excess, ee) being, for example, higher than 30 %, preferably higher than 50 %, and yields of more than 90 % being achievable. The optical yield indicates the ratio of the two stereoisomers formed, which ratio may be, for example, greater than 2:1 and preferably greater than 4:1.

The imines are preferably imines of formula I



- 3 -

which are hydrogenated to form amines of formula II



wherein

$R_3$  is preferably a substituent and wherein

$R_3$  is linear or branched  $C_1$ - $C_{12}$ alkyl, cycloalkyl having from 3 to 8 ring carbon atoms; heterocycloalkyl bonded *via* a carbon atom and having from 3 to 8 ring atoms and 1 or 2 hetero atoms from the group O, S and  $NR_6$ ; a  $C_7$ - $C_{16}$ aralkyl bonded *via* an alkyl carbon atom or  $C_1$ - $C_{12}$ alkyl substituted by the mentioned cycloalkyl or heterocycloalkyl or heteroaryl;

or wherein

$R_3$  is  $C_6$ - $C_{12}$ aryl, or  $C_4$ - $C_{11}$ heteroaryl bonded *via* a ring carbon atom and having 1 or 2 hetero atoms in the ring;  $R_3$  being unsubstituted or substituted by -CN, - $NO_2$ , F, Cl,  $C_1$ - $C_{12}$ alkyl,  $C_1$ - $C_{12}$ alkoxy,  $C_1$ - $C_{12}$ alkylthio,  $C_1$ - $C_6$ haloalkyl, -OH,  $C_6$ - $C_{12}$ -aryl or -aryloxy or -arylthio,  $C_7$ - $C_{16}$ -aralkyl or -aralkoxy or -aralkylthio, secondary amino having from 2 to 24 carbon atoms, -CONR<sub>4</sub>R<sub>5</sub> or by -COOR<sub>4</sub>, and the aryl radicals and the aryl groups in the aralkyl, aralkoxy and aralkylthio in turn being unsubstituted or substituted by -CN, - $NO_2$ , F, Cl,  $C_1$ - $C_4$ -alkyl, -alkoxy or -alkylthio, -OH, -CONR<sub>4</sub>R<sub>5</sub> or by -COOR<sub>4</sub>;  $R_4$  and  $R_5$  are each independently of the other hydrogen,  $C_1$ - $C_{12}$ alkyl, phenyl or benzyl, or  $R_4$  and  $R_5$  together are tetra- or penta-methylene or 3-oxapentylene;

$R_6$  has independently the same meaning as given for  $R_4$ ;

$R_1$  and  $R_2$  are each independently of the other a hydrogen atom,  $C_1$ - $C_{12}$ alkyl or cycloalkyl having from 3 to 8 ring carbon atoms, each of which is unsubstituted or substituted by -OH,  $C_1$ - $C_{12}$ alkoxy, phenoxy, benzyloxy, secondary amino having from 2 to 24 carbon atoms, -CONR<sub>4</sub>R<sub>5</sub> or by -COOR<sub>4</sub>;  $C_6$ - $C_{12}$ aryl or  $C_7$ - $C_{16}$ aralkyl that is unsubstituted or substituted as  $R_3$ , or -CONR<sub>4</sub>R<sub>5</sub> or -COOR<sub>4</sub>, wherein  $R_4$  and  $R_5$  are as defined hereinbefore; or

$R_3$  is as defined hereinbefore and  $R_1$  and  $R_2$  together are alkylene having from 2 to 5 carbon atoms that is optionally interrupted by 1 or 2 -O-, -S- or -NR<sub>6</sub>- radicals, and/or unsubstituted or substituted by =O or as  $R_1$  and  $R_2$  above in the meaning of alkyl, and/or condensed with benzene, pyridine, pyrimidine, furan, thiophene or pyrrole; or

$R_2$  is as defined hereinbefore and  $R_1$  and  $R_3$  together are alkylene having from 2 to 5 carbon atoms that is optionally interrupted by 1 or 2 -O-, -S- or -NR<sub>6</sub>- radicals, and/or unsubstituted or substituted by =O or as  $R_1$  and  $R_2$  above in the meaning of alkyl, and/or condensed with benzene, pyridine, pyrimidine, furan, thiophene or pyrrole.

The radicals  $R_1$ ,  $R_2$  and  $R_3$  may contain one or more chirality centres.

$R_1$ ,  $R_2$  and  $R_3$  can be substituted in any desired positions by identical or different radicals, for example by from 1 to 5, preferably from 1 to 3, substituents.

Suitable substituents for  $R_1$  and  $R_2$  and  $R_3$  are: C<sub>1</sub>-C<sub>12</sub>-, preferably C<sub>1</sub>-C<sub>6</sub>-, and especially C<sub>1</sub>-C<sub>4</sub>-alkyl, -alkoxy or -alkylthio, e.g. methyl, ethyl, propyl, n-, iso- and tert-butyl, the isomers of pentyl, hexyl, octyl, nonyl, decyl, undecyl and dodecyl, and corresponding alkoxy and alkylthio radicals;

C<sub>1</sub>-C<sub>6</sub>-, preferably C<sub>1</sub>-C<sub>4</sub>-haloalkyl having preferably F and Cl as halogen, e.g. trifluoro- or trichloro-methyl, difluorochloromethyl, fluorodichloromethyl, 1,1-difluoroeth-1-yl, 1,1-dichloroeth-1-yl, 1,1,1-trichloro- or 1,1,1-trifluoroeth-2-yl, pentachloroethyl, pentafluoroethyl, 1,1,1-trifluoro-2,2-dichloroethyl, n-perfluoropropyl, iso-perfluoropropyl, n-perfluorobutyl, fluoro- or chloro-methyl, difluoro- or dichloro-methyl, 1-fluoro- or 1-chloro-eth-2-yl or -eth-1-yl, 1-, 2- or 3-fluoro- or 1-, 2- or 3-chloro-prop-1-yl or -prop-2-yl or -prop-3-yl, 1-fluoro- or 1-chloro-but-1-yl, -but-2-yl, -but-3-yl or -but-4-yl, 2,3-dichloro-prop-1-yl, 1-chloro-2-fluoro-prop-3-yl, 2,3-dichlorobut-1-yl;

C<sub>6</sub>-C<sub>12</sub>-aryl, -aryloxy or -arylthio, in which aryl is preferably naphthyl and especially phenyl, C<sub>7</sub>-C<sub>16</sub>-aralkyl, -aralkoxy and -aralkylthio, in which the aryl radical is preferably naphthyl and especially phenyl and the alkylene radical is linear or branched and contains from 1 to 10, preferably from 1 to 6 and especially from 1 to 3, carbon atoms, for example benzyl, naphthylmethyl, 1- or 2-phenyl-eth-1-yl or -eth-2-yl, 1-, 2- or 3-phenyl-prop-1-yl, -prop-2-yl or -prop-3-yl, with benzyl being especially preferred;

the radicals containing the aryl groups mentioned above may in turn be mono- or poly-substituted, for example by C<sub>1</sub>-C<sub>4</sub>-alkyl, -alkoxy or -alkylthio, halogen, -OH, -CONR<sub>4</sub>R<sub>5</sub> or by -COOR<sub>5</sub>, wherein R<sub>4</sub> and R<sub>5</sub> are as defined; examples are methyl, ethyl, n- and isopropyl, butyl, corresponding alkoxy and alkylthio radicals, F, Cl, Br, dimethyl-, methyl-ethyl- and diethyl-carbamoyl and methoxy-, ethoxy-, phenoxy- and benzyloxy-carbonyl;

halogen, preferably F and Cl;

secondary amino having from 2 to 24, preferably from 2 to 12 and especially from 2 to 6 carbon atoms, the secondary amino preferably containing 2 alkyl groups, for example dimethyl-, methylethyl-, diethyl-, methylpropyl-, methyl-n-butyl-, di-n-propyl-, di-n-butyl-, di-n-hexyl-amino;

-CONR<sub>4</sub>R<sub>5</sub>, wherein R<sub>4</sub> and R<sub>5</sub> are each independently of the other C<sub>1</sub>-C<sub>12</sub>-, preferably C<sub>1</sub>-C<sub>6</sub>-, and especially C<sub>1</sub>-C<sub>4</sub>-alkyl, or R<sub>4</sub> and R<sub>5</sub> together are tetra- or penta-methylene or 3-oxapentylene, the alkyl being linear or branched, e.g. dimethyl-, methylethyl-, diethyl-, methyl-n-propyl-, ethyl-n-propyl-, di-n-propyl-, methyl-n-butyl-, ethyl-n-butyl-, n-propyl-n-butyl- and di-n-butyl-carbamoyl;

-COOR<sub>4</sub>, wherein R<sub>4</sub> is C<sub>1</sub>-C<sub>12</sub>-, preferably C<sub>1</sub>-C<sub>6</sub>-alkyl, which may be linear or branched, e.g. methyl, ethyl, n- and iso-propyl, n-, iso- and tert-butyl, and the isomers of pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl and dodecyl.

R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> may contain especially functional groups, such as keto groups, -CN, -NO<sub>2</sub>, carbon double bonds, N-O-, aromatic halogen groups and amide groups.

R<sub>1</sub> and R<sub>2</sub> as heteroaryl are preferably a 5- or 6-membered ring having 1 or 2 identical or different hetero atoms, especially O, S or N, which contains preferably 4 or 5 carbon atoms and can be condensed with benzene. Examples of heteroaromatics from which R<sub>1</sub> can be derived are furan, pyrrole, thiophene, pyridine, pyrimidine, indole and quinoline.

R<sub>1</sub> and R<sub>2</sub> as heteroaryl-substituted alkyl are derived preferably from a 5- or 6-membered ring having 1 or 2 identical or different hetero atoms, especially O, S or N, which contains preferably 4 or 5 carbon atoms and can be condensed with benzene. Examples of heteroaromatics are furan, pyrrole, thiophene, pyridine, pyrimidine, indole and quinoline.

R<sub>1</sub> and R<sub>2</sub> as heterocycloalkyl or as heterocycloalkyl-substituted alkyl contain preferably from 4 to 6 ring atoms and 1 or 2 identical or different hetero atoms from the group O, S and NR<sub>6</sub>. It can be condensed with benzene. It may be derived, for example, from pyrrolidine, tetrahydrofuran, tetrahydrothiophene, indane, pyrazolidine, oxazolidine, piperidine, piperazine or morpholine.

R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> as alkyl are preferably unsubstituted or substituted C<sub>1</sub>-C<sub>6</sub>-, especially

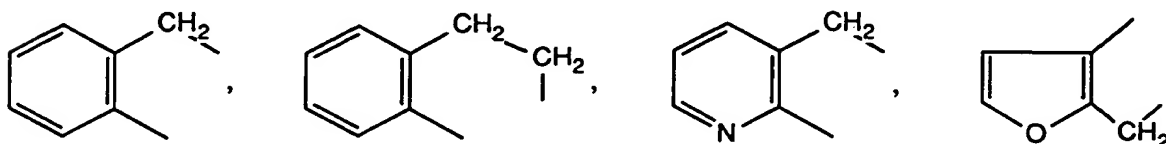
C<sub>1</sub>-C<sub>4</sub>-alkyl, which may be linear or branched. Examples are methyl, ethyl, iso- and n-propyl, iso-, n- and tert-butyl, the isomers of pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl and dodecyl.

R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> as unsubstituted or substituted cycloalkyl contain preferably from 3 to 6, especially 5 or 6, ring carbon atoms. Examples are cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl and cyclooctyl.

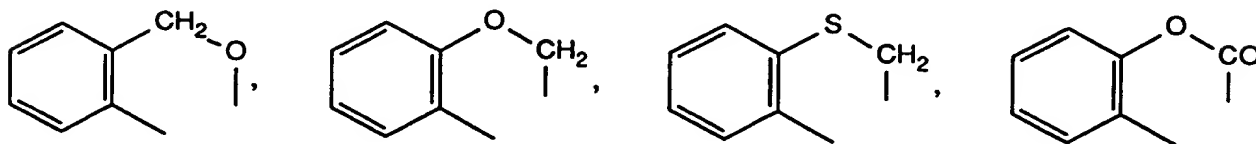
R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> as aryl are preferably unsubstituted or substituted naphthyl and especially phenyl. R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> as aralkyl are preferably unsubstituted or substituted phenylalkyl having from 1 to 10, preferably from 1 to 6 and especially from 1 to 4 carbon atoms in the alkylene, the alkylene being linear or branched. Examples are especially benzyl, and 1-phenyleth-1-yl, 2-phenyleth-1-yl, 1-phenylprop-1-yl, 1-phenylprop-2-yl, 1-phenylprop-3-yl, 2-phenylprop-1-yl, 2-phenylprop-2-yl and 1-phenylbut-4-yl.

In R<sub>2</sub> and R<sub>3</sub> as -CONR<sub>4</sub>R<sub>5</sub> and -COOR<sub>4</sub>, R<sub>4</sub> and R<sub>5</sub> are preferably C<sub>1</sub>-C<sub>6</sub>-, especially C<sub>1</sub>-C<sub>4</sub>-alkyl, or R<sub>4</sub> and R<sub>5</sub> together are tetramethylene, pentamethylene or 3-oxapentylene. Examples of alkyl are mentioned hereinbefore.

R<sub>1</sub> and R<sub>2</sub> together or R<sub>1</sub> and R<sub>3</sub> together as alkylene are preferably interrupted by 1 -O-, -S- or -NR<sub>6</sub>-, preferably -O-. R<sub>1</sub> and R<sub>2</sub> together or R<sub>1</sub> and R<sub>3</sub> together form, with the carbon atom or with the -N=C group to which they are bonded, respectively, preferably a 5- or 6-membered ring. For the substituents the preferences mentioned hereinbefore apply. As condensed alkylene, R<sub>1</sub> and R<sub>2</sub> together or R<sub>1</sub> and R<sub>3</sub> together are preferably alkylene condensed with benzene or pyridine. Examples of alkylene are: ethylene, 1,2- or 1,3-propylene, 1,2-, 1,3- or 1,4-butylene, 1,5-pentylene and 1,6-hexylene. Examples of interrupted or =O-substituted alkylene are 2-oxa-1,3-propylene, 2-oxa-1,4-butylene, 2-oxa- or 3-oxa-1,5-pentylene, 3-thia-1,5-pentylene, 2-thia-1,4-butylene, 2-thia-1,3-propylene, 2-methylimino-1,3-propylene, 2-ethylimino-1,4-butylene, 2- or 3-methylimino-1,5-pentylene, 1-oxo-2-oxa-1,3-propylene, 1-oxo-2-oxa-1,4-butylene, 2-oxo-3-oxa-1,4-butylene, 1-oxa-2-oxo-1,5-pentylene. Examples of condensed alkylene are:



Examples of condensed and interrupted and unsubstituted or =O-substituted alkylene are:

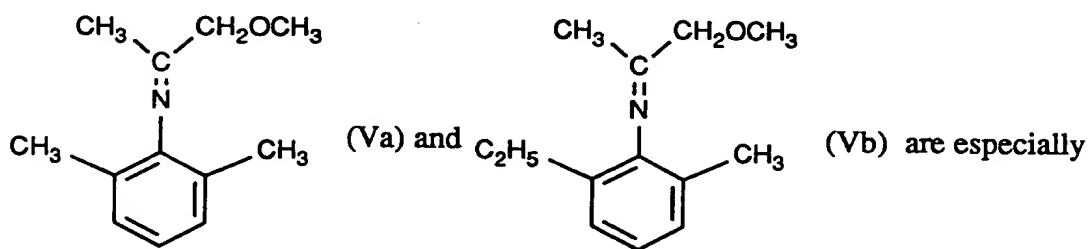


$R_4$  and  $R_5$  are preferably each independently of the other hydrogen,  $C_1$ - $C_4$ alkyl, phenyl or benzyl.  $R_6$  is preferably hydrogen or  $C_1$ - $C_4$ alkyl.

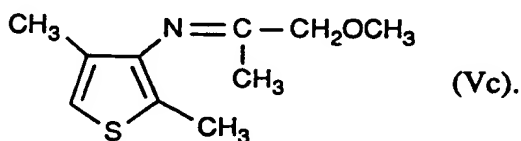
A further preferred group is formed by prochiral imines in which in formula I  $R_1$ ,  $R_2$  and  $R_3$  are each different from the others and are not hydrogen.

In an especially preferred group, in formula I  $R_3$  is 2,6-di- $C_1$ - $C_4$ alkylphen-1-yl and especially 2,6-dimethylphen-1-yl or 2-methyl-6-ethylphen-1-yl,  $R_1$  is  $C_1$ - $C_4$ alkyl and especially ethyl or methyl, and  $R_2$  is  $C_1$ - $C_4$ alkyl,  $C_1$ - $C_4$ alkoxymethyl or  $C_1$ - $C_4$ alkoxyethyl, and especially methoxymethyl.

Of those compounds, imines of formulae



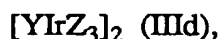
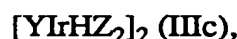
important, as is the imine of the formula



Imines of formula I are known or they can be prepared in accordance with known processes from aldehydes or ketones and primary amines.

The iridium catalysts are preferably homogeneous catalysts that are substantially soluble

in the reaction medium. The term "catalyst" also includes catalyst precursors that are converted into an active catalyst species at the beginning of a hydrogenation. The catalysts preferably correspond to the formulae III, IIIa, IIIb, IIIc and IIId,



wherein X is two olefin ligands or a diene ligand, Y is a ditertiary diphosphine

(a) the phosphine groups of which are bonded to different carbon atoms of a carbon chain having from 2 to 4 carbon atoms, or

(b) the phosphine groups of which are either bonded directly or *via* a bridge group  $-\text{CR}_a\text{R}_b-$  in the ortho positions of a cyclopentadienyl ring or are each bonded to a cyclopentadienyl ring of a ferrocenyl, or

(c) one phosphine group of which is bonded to a carbon chain having 2 or 3 carbon atoms and the other phosphine group of which is bonded to an oxygen atom or a nitrogen atom bonded terminally to that carbon chain, or

(d) the phosphine groups of which are bonded to the two oxygen atoms or nitrogen atoms bonded terminally to a  $\text{C}_2$ -carbon chain;

with the result that in the cases of (a), (b), (c) and (d) a 5-, 6- or 7-membered ring is formed together with the Ir atom, the radicals Z are each independently of the other(s) Cl, Br or I,  $\text{A}^{\ominus}$  is the anion of an oxy or complex acid, and  $\text{M}^{\oplus}$  is an alkali metal cation or quaternary ammonium, and  $\text{R}_a$  and  $\text{R}_b$  are each independently of the other hydrogen,  $\text{C}_1$ - $\text{C}_8$ alkyl,  $\text{C}_1$ - $\text{C}_4$ fluoroalkyl, phenyl or benzyl or are phenyl or benzyl having from 1 to 3  $\text{C}_1$ - $\text{C}_4$ alkyl or  $\text{C}_1$ - $\text{C}_4$ alkoxy substituents.  $\text{R}_b$  is preferably hydrogen.  $\text{R}_a$  is preferably  $\text{C}_1$ - $\text{C}_4$ alkyl and especially methyl.

The diphosphine Y contains preferably at least one chiral carbon atom and is especially an optically pure stereoisomer (enantiomer or diastereoisomer), or a pair of diastereoisomers, since the use of catalysts containing those ligands leads to optical induction in asymmetric hydrogenation reactions.

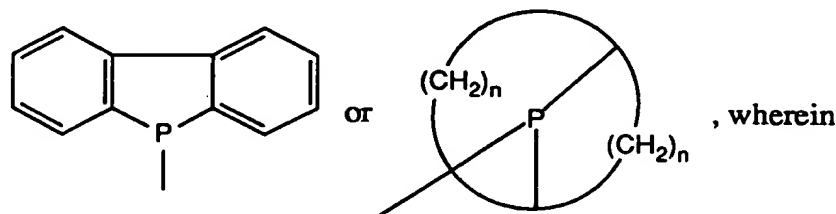
X as an olefin ligand may be a branched or, preferably, linear  $\text{C}_2$ - $\text{C}_{12}$ alkylene, especially  $\text{C}_2$ - $\text{C}_6$ alkylene. Some examples are dodecylene, decylene, octylene, 1-, 2- or 3-hexene, 1-,



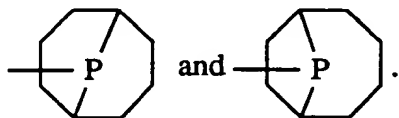
2- or 3-pentene, 1- or 2-butene, propene and ethene. X as a diene ligand may be open-chain or cyclic dienes having from 4 to 12, preferably from 5 to 8, carbon atoms, the diene groups preferably being separated by one or two saturated carbon atoms. Some examples are butadiene, pentadiene, hexadiene, heptadiene, octadiene, decadiene, dodecadiene, cyclopentadiene, cyclohexadiene, cycloheptadiene, cyclooctadiene and bridged cyclo-dienes such as norbornadiene and bicyclo-2,2,2-octadiene. Hexadiene, cyclooctadiene and norbornadiene are preferred.

The phosphine groups contain preferably two identical or different, preferably identical, unsubstituted or substituted hydrocarbon radicals having from 1 to 20, especially from 1 to 12 carbon atoms. Preference is given to diphosphines wherein the secondary phosphine groups contain two identical or different radicals from the following group: linear or branched  $C_1$ - $C_{12}$ alkyl; unsubstituted or  $C_1$ - $C_6$ alkyl- or  $C_1$ - $C_6$ alkoxy-substituted  $C_5$ - $C_{12}$ -cycloalkyl,  $C_5$ - $C_{12}$ cycloalkyl- $CH_2$ -, phenyl or benzyl; and phenyl or benzyl substituted by halogen (e.g. F, Cl or Br),  $C_1$ - $C_6$ haloalkyl,  $(C_1$ - $C_{12}$ alkyl) $_3$ Si,  $(C_6H_5)_3$ Si,  $C_1$ - $C_6$ haloalkoxy (e.g. trifluoromethoxy),  $-NH_2$ , phenyl $_2$ N-, benzyl $_2$ N-, morpholinyl, piperidinyl, pyrrolidinyl,  $(C_1$ - $C_{12}$ alkyl) $_2$ N-, -ammonium- $X_1^\ominus$ ,  $-SO_3M_1$ ,  $-CO_2M_1$ ,  $-PO_3M_1$  or by  $-COO$ - $C_1$ - $C_6$ -alkyl (e.g.  $-COOCH_3$ ), wherein  $M_1$  is an alkali metal or hydrogen and  $X_1^\ominus$  is the anion of a monobasic acid.  $M_1$  is preferably H, Li, Na or K.  $A_1^\ominus$ , as the anion of a monobasic acid, is preferably  $Cl^\ominus$ ,  $Br^\ominus$  or the anion of a carboxylic acid, for example formate, acetate, trichloroacetate or trifluoroacetate.

A secondary phosphine group may also be a radical of the formula



m and n are each independently of the other an integer from 2 to 10, and the sum of m+n is from 4 to 12, especially from 5 to 8. Examples thereof are [3.3.1]- and [4.2.1]-phosbyl of the formulae



Examples of alkyl that preferably contains from 1 to 6 carbon atoms are methyl, ethyl, n-propyl, isopropyl, n-, iso- and tert-butyl and the isomers of pentyl and hexyl. Examples of unsubstituted or alkyl-substituted cycloalkyl are cyclopentyl, cyclohexyl, methyl- or ethyl-cyclohexyl and dimethylcyclohexyl. Examples of alkyl-, alkoxy- or haloalkoxy-substituted phenyl and benzyl are methylphenyl, dimethylphenyl, trimethylphenyl, ethylphenyl, methylbenzyl, methoxyphenyl, dimethoxyphenyl, trifluoromethylphenyl, bis-trifluoromethylphenyl, tris-trifluoromethylphenyl, trifluoromethoxyphenyl and bis-trifluoromethoxyphenyl. Preferred phosphine groups are those that contain identical or different, preferably identical, radicals from the group C<sub>1</sub>-C<sub>6</sub>alkyl; cyclopentyl and cyclohexyl that are unsubstituted or have from 1 to 3 C<sub>1</sub>-C<sub>4</sub>alkyl or C<sub>1</sub>-C<sub>4</sub>alkoxy substituents, benzyl and, especially, phenyl that is unsubstituted or has from 1 to 3 C<sub>1</sub>-C<sub>4</sub>alkyl, C<sub>1</sub>-C<sub>4</sub>alkoxy, F, Cl, C<sub>1</sub>-C<sub>4</sub>fluoroalkyl or C<sub>1</sub>-C<sub>4</sub>fluoroalkoxy substituents.

Y as a diphosphine is preferably of formula IV, IVa, IVb, IVc or IVd,

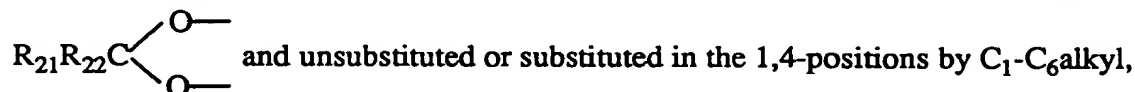


wherein

R<sub>7</sub>, R<sub>8</sub>, R<sub>10</sub> and R<sub>11</sub> are each independently of the others a hydrocarbon radical having from 1 to 20 carbon atoms that is unsubstituted or substituted by C<sub>1</sub>-C<sub>6</sub>alkyl, C<sub>1</sub>-C<sub>6</sub>alkoxy, halogen, C<sub>1</sub>-C<sub>6</sub>haloalkyl, (C<sub>1</sub>-C<sub>12</sub>alkyl)<sub>3</sub>Si, (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>Si, C<sub>1</sub>-C<sub>6</sub>haloalkoxy, -NH<sub>2</sub>, phenyl<sub>2</sub>N-, benzyl<sub>2</sub>N-, morpholinyl, piperidinyl, pyrrolidinyl, (C<sub>1</sub>-C<sub>12</sub>alkyl)<sub>2</sub>N-, -ammonium-X<sub>1</sub><sup>⊖</sup>, -SO<sub>3</sub>M<sub>1</sub>, -CO<sub>2</sub>M<sub>1</sub>, -PO<sub>3</sub>M<sub>1</sub> or by -COO-C<sub>1</sub>-C<sub>6</sub>alkyl, wherein M<sub>1</sub> is an alkali metal or hydrogen and X<sub>1</sub><sup>⊖</sup> is the anion of a monobasic acid;

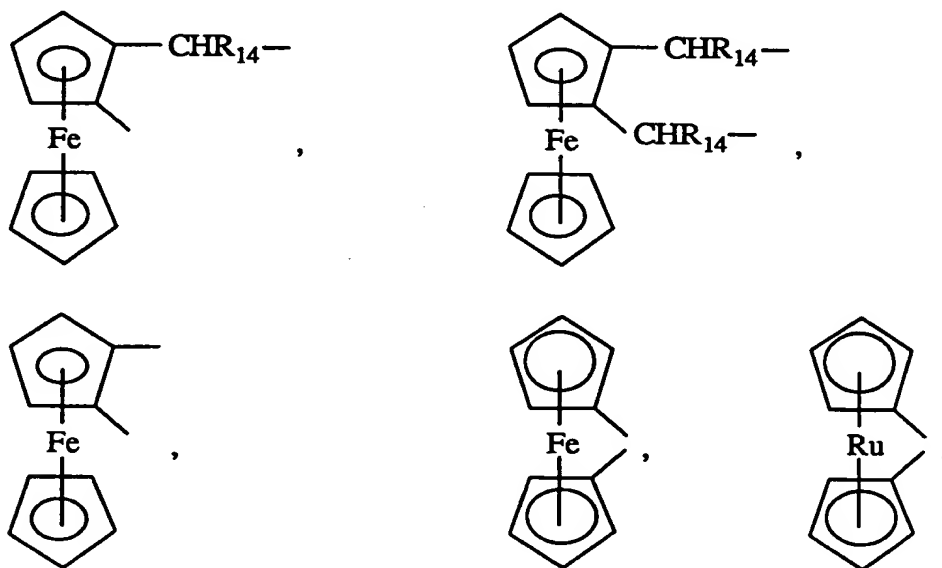
R<sub>9</sub> is linear C<sub>2</sub>-C<sub>4</sub>alkylene that is unsubstituted or substituted by C<sub>1</sub>-C<sub>6</sub>alkyl, C<sub>5</sub>- or C<sub>6</sub>-cycloalkyl, phenyl, naphthyl or by benzyl; 1,2- or 1,3-cycloalkylene or -cycloalkenyl-

ene, -bicycloalkylene or -bicycloalkenylene having from 4 to 10 carbon atoms, each of which is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl, phenyl or by benzyl; 1,2- or 1,3-cycloalkylene or -cycloalkenylene, -bicycloalkylene or -bicycloalkenylene having from 4 to 10 carbon atoms, each of which is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl, phenyl or by benzyl, and in the 1- and/or 2-positions or in the 3-position of which methylene or  $C_2$ - $C_4$ alkylidene is bonded; 1,4-butylene substituted in the 2,3-positions by

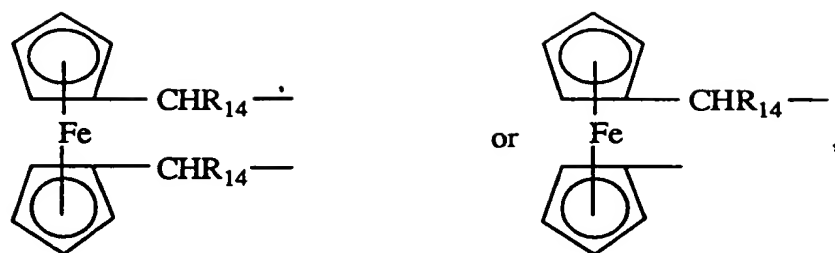


phenyl or by benzyl, wherein  $R_{21}$  and  $R_{22}$  are each independently of the other hydrogen,  $C_1$ - $C_6$ alkyl, phenyl or benzyl; 3,4- or 2,4-pyrrolidinylene or 2-methylene-pyrrolidin-4-yl the nitrogen atom of which is substituted by hydrogen,  $C_1$ - $C_{12}$ alkyl, phenyl, benzyl,  $C_1$ - $C_{12}$ alkoxycarbonyl,  $C_1$ - $C_8$ acyl or by  $C_1$ - $C_{12}$ alkylaminocarbonyl; or 1,2-phenylene, 2-benzylene, 1,2-xylylene, 1,8-naphthylene, 2,2'-dinaphthylene or 2,2'-diphenylene, each of which is unsubstituted or substituted by  $C_1$ - $C_4$ alkyl;

or  $R_9$  is a radical of the formula



- 12 -



wherein  $R_{14}$  is hydrogen,  $C_1$ - $C_8$ alkyl,  $C_1$ - $C_4$ fluoroalkyl, phenyl or phenyl having from 1 to 3  $C_1$ - $C_4$ alkyl or  $C_1$ - $C_4$ alkoxy substituents;

$R_{12}$  is linear  $C_2$ - or  $C_3$ -alkylene that is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl,  $C_5$ - or  $C_6$ -cycloalkyl, phenyl, naphthyl or by benzyl; 1,2- or 1,3-cycloalkylene or -cycloalkenylene, -bicycloalkylene or -bicycloalkenylene having from 4 to 10 carbon atoms, each of which is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl, phenyl or by benzyl; or 1,2- or 1,3-cycloalkylene or -cycloalkenylene, -bicycloalkylene or -bicycloalkenylene having from 4 to 10 carbon atoms, each of which is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl, phenyl or by benzyl, and in the 1- and/or 2-positions or in the 3-position of which methylene or  $C_2$ - $C_4$ alkylidene is bonded; 3,4- or 2,4-pyrrolidinylene or 3-methylene-pyrrolidin-4-yl the nitrogen atom of which is substituted by hydrogen,  $C_1$ - $C_{12}$ alkyl, phenyl, benzyl,  $C_1$ - $C_{12}$ alkoxycarbonyl,  $C_1$ - $C_8$ acyl or by  $C_1$ - $C_{12}$ alkylaminocarbonyl; or 1,2-phenylene, 2-benzylene, 1,2-, 2,3- or 1,8-naphthylene, each of which is unsubstituted or substituted by  $C_1$ - $C_4$ alkyl; and

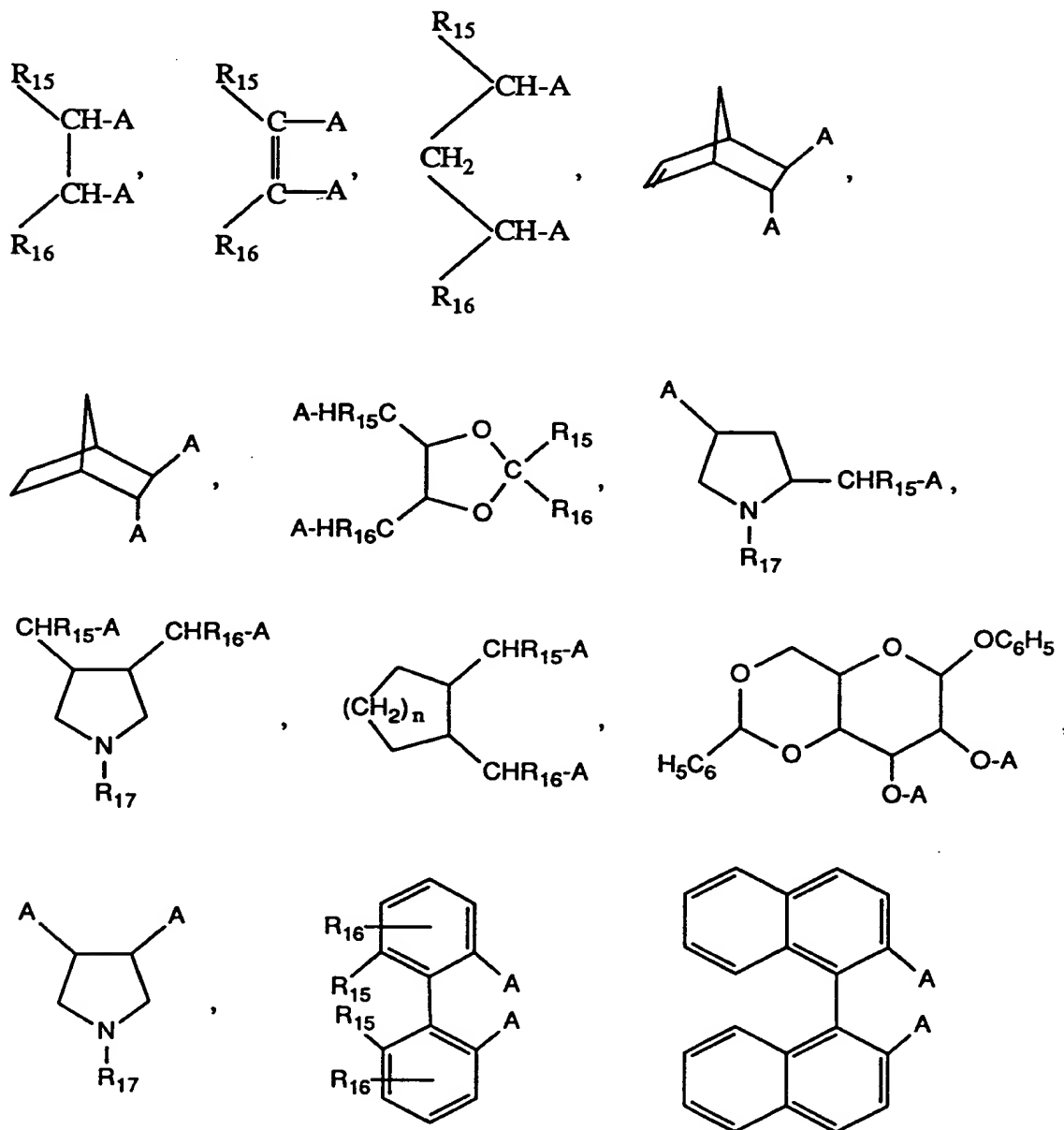
$R_{13}$  is linear  $C_2$ alkylene that is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl,  $C_5$ - or  $C_6$ -cycloalkyl, phenyl, naphthyl or by benzyl; 1,2-cycloalkylene or -cycloalkenylene, -bicycloalkylene or -bicycloalkenylene having from 4 to 10 carbon atoms, each of which is unsubstituted or substituted by  $C_1$ - $C_6$ alkyl, phenyl or by benzyl; 3,4-pyrrolidinylene the nitrogen atom of which is substituted by hydrogen,  $C_1$ - $C_{12}$ alkyl, phenyl, benzyl,  $C_1$ - $C_{12}$ alkoxycarbonyl,  $C_1$ - $C_8$ acyl or by  $C_1$ - $C_{12}$ alkylaminocarbonyl; or 1,2-phenylene that is unsubstituted or substituted by  $C_1$ - $C_4$ alkyl, or is a radical, less two hydroxy groups in the ortho positions, of a mono- or di-saccharide, and

$R_c$  is hydrogen,  $C_1$ - $C_4$ alkyl, phenyl or benzyl.

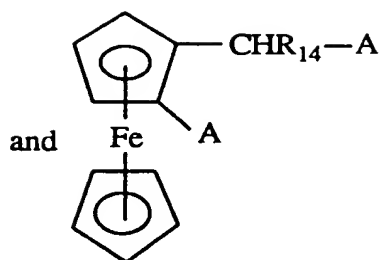
$R_7$ ,  $R_8$ ,  $R_{10}$  and  $R_{11}$  are preferably identical or different, preferably identical, radicals from the following group:  $C_1$ - $C_6$ alkyl; cyclopentyl and cyclohexyl that are unsubstituted or

have from 1 to 3 C<sub>1</sub>-C<sub>4</sub>alkyl or C<sub>1</sub>-C<sub>4</sub>alkoxy substituents, benzyl and, especially, phenyl that is unsubstituted or has from 1 to 3 C<sub>1</sub>-C<sub>4</sub>alkyl, C<sub>1</sub>-C<sub>4</sub>alkoxy, F, Cl, C<sub>1</sub>-C<sub>4</sub>fluoroalkyl or C<sub>1</sub>-C<sub>4</sub>fluoroalkoxy substituents.

A preferred subgroup of diphosphines Y is formed by those of the formulae



- 14 -



wherein

$R_{15}$  and  $R_{16}$  are each independently of the other hydrogen,  $C_1$ - $C_4$ alkyl, phenyl, benzyl, or phenyl or benzyl having from 1 to 3  $C_1$ - $C_4$ alkyl or  $C_1$ - $C_4$ alkoxy substituents,

$R_{14}$  is hydrogen,  $C_1$ - $C_4$ alkyl, phenyl, benzyl, or phenyl or benzyl having from 1 to 3  $C_1$ - $C_4$ alkyl or  $C_1$ - $C_4$ alkoxy substituents,

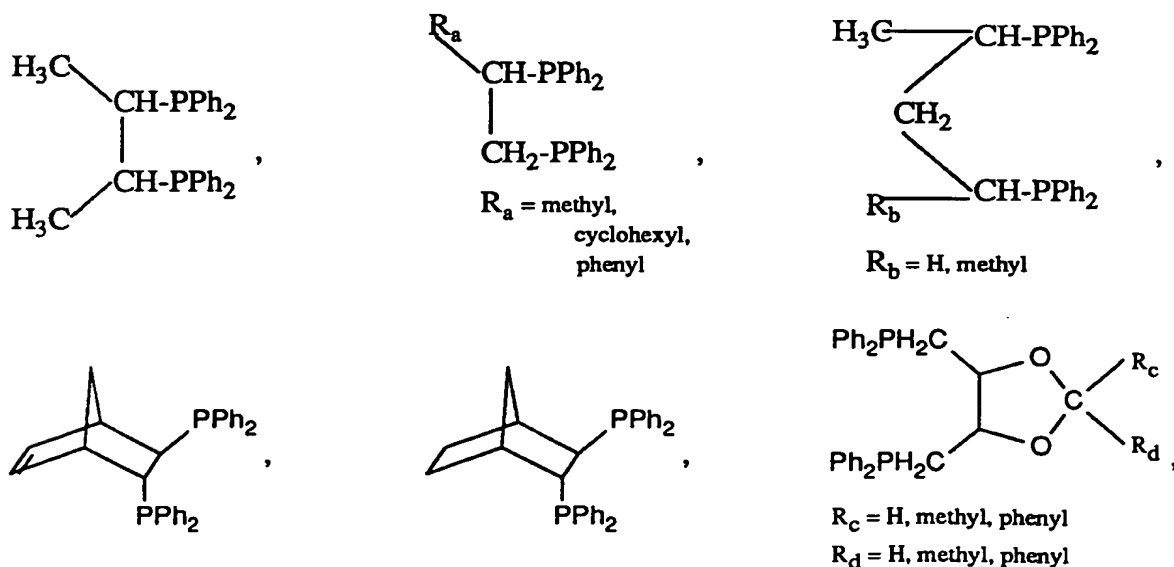
$R_{17}$  is hydrogen,  $C_1$ - $C_4$ alkyl, phenyl, benzyl,  $C_1$ - $C_6$ alkoxy-CO-,  $C_1$ - $C_6$ alkyl-CO-, phenyl-CO-, naphthyl-CO- or  $C_1$ - $C_4$ alkylNH-CO-,

A may be identical or different groups  $-PR_2$ , wherein R is  $C_1$ - $C_6$ alkyl, cyclohexyl, phenyl, benzyl, or phenyl or benzyl having from 1 to 3  $C_1$ - $C_4$ alkyl,  $C_1$ - $C_4$ alkoxy,  $-CF_3$  or partially or fully fluorinated  $C_1$ - $C_4$ alkoxy substituents, and

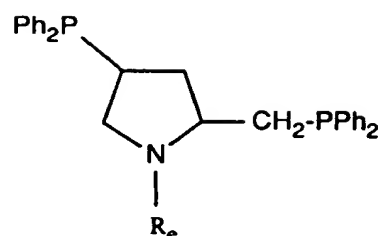
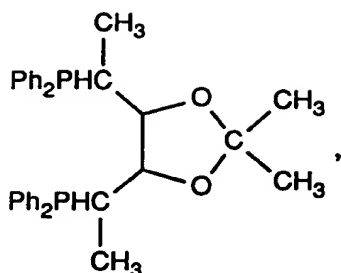
n is 0, 1 or 2.

Of those diphosphines, chirally substituted compounds are especially preferred.

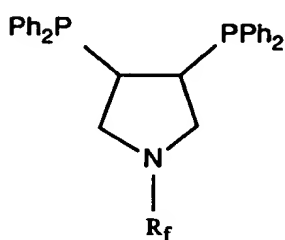
Some preferred examples of diphosphines Y are as follows (Ph is phenyl):



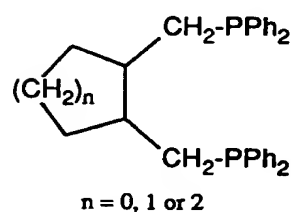
- 15 -



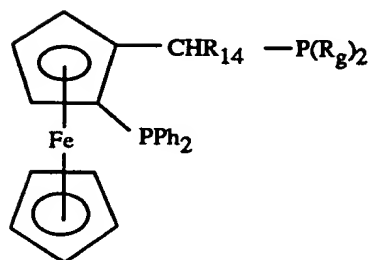
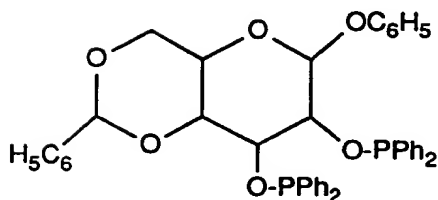
$R_e = -\text{CO}_2\text{-tert-butyl}, -\text{CO-tert-butyl}, \text{H},$   
 $-\text{CO-phenyl}, -\text{CO-NH-C}_1\text{-C}_4\text{alkyl}$



$R_f = \text{C}_1\text{-C}_4\text{alkyl}, \text{benzyl}$



$n = 0, 1 \text{ or } 2$



$R_{14} = \text{C}_1\text{-C}_4\text{alkyl}, \text{especially methyl},$   
 $R_g = \text{phenyl or cyclohexyl that is}$   
 unsubstituted or has from 1  
 to 3 methyl,  $-\text{CF}_3$  or methoxy  
 substituents

Suitable diphosphines and diphosphinites have been described, for example, by H.B. Kagan in *Chiral Ligands for Asymmetric Catalysis, Asymmetric Synthesis*, Volume 5, pp. 13-23, Academic Press, Inc., N.Y. (1985). The preparation of ferrocenyl diphosphine ligands is described, for example, in EP-A-0 564 406 and by T. Hayashi et al. in *Bull. Chem. Soc. Jpn.*, 53, pages 1136-1151.

$A^\ominus$  in formula IIIa can be derived from inorganic or organic oxy acids. Examples of such acids are  $\text{H}_2\text{SO}_4$ ,  $\text{HClO}_4$ ,  $\text{HClO}_3$ ,  $\text{HBrO}_4$ ,  $\text{HIO}_4$ ,  $\text{HNO}_3$ ,  $\text{H}_3\text{PO}_3$ ,  $\text{H}_3\text{PO}_4$ ,  $\text{CF}_3\text{SO}_3\text{H}$ ,  $\text{C}_6\text{H}_5\text{SO}_3\text{H}$ ,  $\text{CF}_3\text{COOH}$  and  $\text{CCl}_3\text{COOH}$ . Complex acids from which  $A^\ominus$  can be derived

are, for example, the halo complex acids of the elements B, P, As, Sb and Bi. Preferred examples of  $A^{\ominus}$  in formula IIIa are  $ClO_4^{\ominus}$ ,  $CF_3SO_3^{\ominus}$ ,  $BF_4^{\ominus}$ ,  $B(phenyl)_4^{\ominus}$ ,  $PF_6^{\ominus}$ ,  $SbCl_6^{\ominus}$ ,  $AsF_6^{\ominus}$  and  $SbF_6^{\ominus}$ .

When  $M^{\oplus}$  in formula IIIb is an alkali metal cation, it may be, for example, a Li, Na, K, Rb or Cs cation. When  $M^{\oplus}$  is quaternary ammonium, it may contain a total of from 4 to 40, preferably from 4 to 24, carbon atoms.  $M^{\oplus}$  may correspond to the formula  $phenyl-N^{\oplus}(C_1-C_6alkyl)_3$ ,  $benzylN^{\oplus}(C_1-C_6alkyl)_3$  or  $(C_1-C_6alkyl)_4N^{\oplus}$ .  $M^{\oplus}$  in formula IIIb is preferably  $Li^{\oplus}$ ,  $Na^{\oplus}$  or  $K^{\oplus}$  or  $(C_1-C_6alkyl)_4N^{\oplus}$ .

Z in formula III is preferably Br or Cl and especially Cl. Z in formula IIb is preferably Br or I and Z in formulae IIIc and IIId is preferably I.

Especially suitable diphosphine ligands which can preferably be used in catalysts of formula (III) are, for example:

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dipropyl-aminophenyl)phosphine

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-diisopropyl-4-N,N-dimethyl-aminophenyl)phosphine

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-diisopropyl-4-N,N-dibenzyl-aminophenyl)phosphine

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dibenzyl-aminophenyl)phosphine

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-(1'-pyrrolo)-phenyl)phosphine

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dipentyl-aminophenyl)phosphine

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dimethyl-aminophenyl)phosphine

1,4-bis(diphenylphosphino)butane

{(R)-1-[(S)-2-di(4-methoxyphenyl)phosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dimethylaminophenyl)phosphine and preferably

{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-phenyl)phosphine.

The preparation of the catalysts is known *per se* and is described, for example, in US-A-4 994 615, US-A-5 011 995, US-A-5 112 999 and EP-A-0 564 406. The preparation



of the catalysts of formula III can be carried out, for example, by reacting a diiridium complex of the formula  $[\text{IrXZ}]_2$  with a diphosphine Y. The iridium catalysts can be added to the reaction mixture as isolated compounds. It has proved advantageous, however, to produce the catalyst *in situ* with or without a solvent prior to the reaction and to add optionally a portion or all of the acid and of an ammonium or alkali metal halide.

The iridium catalysts are preferably used in amounts of from 0.0001 to 10 mol %, especially from 0.001 to 10 mol %, and more especially from 0.01 to 5 mol %, based on the imine.

The molar ratio of the imine to the iridium catalyst may be, for example, from 5 000 000 to 10, especially from 2 000 000 to 20, more preferably from 1 000 000 to 20, and more especially from 500 000 to 100.

The process is carried out preferably at a temperature of from -20 to 100°C, especially from 0 to 80°C and more especially from 10 to 70°C, and preferably at a hydrogen pressure of  $2 \times 10^5$  to  $1.5 \times 10^7$  Pa (5 to 150 bar), especially  $10^6$  to  $10^7$  Pa (10 to 100 bar).

The chlorides, bromides and iodides employed are preferably used in concentrations of from 0.01 to 500 mmol/l, especially from 0.01 to 50 mmol/l, based on the volume of the reaction mixture.

The process according to the invention comprises the additional concomitant use of an ammonium or metal chloride, bromide or iodide. The chlorides, bromides and iodides are used preferably in amounts of from 0.01 to 200 mol %, especially from 0.05 to 100 mol % and more especially from 0.5 to 50 mol %, based on the iridium catalyst. The iodides are preferred. Ammonium is preferably tetraalkylammonium having from 1 to 6 carbon atoms in the alkyl groups, and the metal is preferably sodium, lithium or potassium. Special preference is given to tetrabutylammonium iodide and sodium.

Provided that they are soluble in the reaction mixture and provided that oxidation reactions with other reactants can be ruled out, virtually any metal chlorides, bromides and iodides, that is to say those of the main groups and sub-groups of the Periodic Table of the Elements, can be used in the process according to the invention.

The reaction can be carried out in the absence or in the presence of solvents. Suitable

solvents, which can be used alone or as a mixture of solvents, are especially aprotic solvents. Examples are:

aliphatic and aromatic hydrocarbons, such as pentane, hexane, cyclohexane, methylcyclohexane, benzene, toluene and xylene; ethers, such as diethyl ether, diethylene glycol dimethyl ether, tetrahydrofuran and dioxane; halogenated hydrocarbons, such as methylene chloride, chloroform, 1,1,2,2-tetrachloroethane and chlorobenzene; esters and lactones, such as ethyl acetate, butyrolactone and valerolactone; acid amides and lactams, such as dimethylformamide, dimethylacetamide and N-methylpyrrolidone, and ketones, such as acetone, dibutyl ketone, methyl isobutyl ketone and methoxyacetone.

The process according to the invention further comprises the additional concomitant use of an acid. It may be an inorganic or, preferably, an organic acid. The acid is preferably used in at least the same molar amount as the iridium catalyst (equivalent to catalytic amounts) and can also be used in excess. The excess may even consist in the use of the acid as solvent. Preferably from 0.001 to 50, in particular from 0.1 to 50 % by weight of acid is used, based on the amine. In many cases it can be advantageous to use anhydrous acids.

Some examples of inorganic acids are  $\text{H}_2\text{SO}_4$ , highly concentrated sulfuric acid (oleum),  $\text{H}_3\text{PO}_4$ , orthophosphoric acid, HF, HCl, HBr, HI,  $\text{HClO}_4$ ,  $\text{HBF}_4$ ,  $\text{HPF}_6$ ,  $\text{HAsF}_6$ ,  $\text{HSbCl}_6$ ,  $\text{HSbF}_6$  and  $\text{HB}(\text{phenyl})_4$ .  $\text{H}_2\text{SO}_4$  is particularly preferred.

Examples of organic acids are aliphatic or aromatic, optionally halogenated (fluorinated or chlorinated) carboxylic acids, sulfonic acids, phosphorus(V) acids (for example phosphonic acids, phosphonous acids) having preferably from 1 to 20, especially from 1 to 12 and more especially from 1 to 6, carbon atoms. Examples are formic acid, acetic acid, propionic acid, butyric acid, benzoic acid, phenylacetic acid, cyclohexanecarboxylic acid, chloro- or fluoro-acetic acid, dichloro- or difluoro-acetic acid, trichloro- or trifluoro-acetic acid, chlorobenzoic acid, methanesulfonic acid, ethanesulfonic acid, benzenesulfonic acid, p-toluenesulfonic acid, chlorobenzenesulfonic acid, trifluoromethanesulfonic acid, methylphosphonic acid and phenylphosphonic acid. Preferred acids are acetic acid, propionic acid, trifluoroacetic acid, methanesulfonic acid and chloroacetic acid.

It is also possible for acidic ion exchangers of an inorganic or organic nature to be used as the acids.

In detail, the process according to the invention can be carried out by first preparing the catalyst by dissolving, for example,  $(\text{Ir-dieneCl})_2$  in a solvent or an acid or both, adding a diphosphine and then an alkali metal or ammonium halide and stirring the mixture.

$(\text{Ir-dieneCl})_2$  can also be used in solid form. A solution of imines is added to that catalyst solution (or *vice versa*) and, in an autoclave, hydrogen pressure is applied, thus removing the protective gas that is advantageously used. It is advantageous to ensure that the catalyst solution stands for only a short time, and to carry out the hydrogenation of the imines as soon as possible after the preparation of the catalyst. The reaction mixture is heated, if desired, and then hydrogenated. Where appropriate, when the reaction has ceased the reaction mixture is cooled and the autoclave is depressurised. The reaction mixture can be removed from the autoclave under pressure with nitrogen and the hydrogenated organic compound can be isolated and purified in a manner known *per se*, for example by precipitation, extraction or distillation.

In the case of the hydrogenation of aldimines and ketimines, the aldimines and ketimines can also be formed *in situ* before or during the hydrogenation. In a preferred form, an amine and an aldehyde or a ketone are mixed together and added to the catalyst solution and the aldimine or ketimine formed *in situ* is hydrogenated. It is also possible, however, to use an amine, a ketone or an aldehyde together with the catalyst as the initial batch and to add the ketone or the aldehyde or the amine thereto, either all at once or in metered amounts.

The hydrogenation can be carried out continuously or batchwise in various types of reactor. Preference is given to those reactors which allow comparatively good intermixing and good removal of heat, such as, for example, loop reactors. That type of reactor has proved to be especially satisfactory when small amounts of catalyst are used.

The process according to the invention yields the corresponding amines in short reaction times while having chemically a high degree of conversion, with surprisingly good optical yields (ee) of 70 % or more being obtained even at relatively high temperatures of more than 50°C, and even with high molar ratios of imine to catalyst.

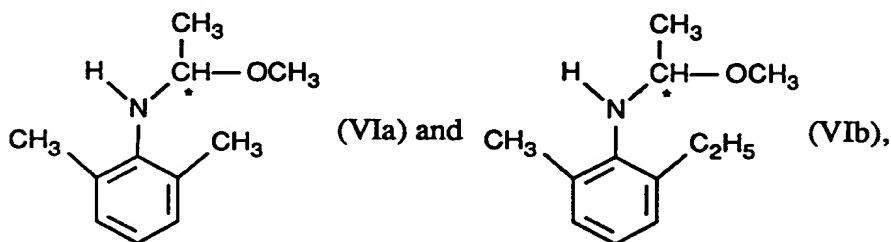
The hydrogenated organic compounds that can be prepared in accordance with the invention, for example the amines, are biologically active substances or are intermediates for the preparation of such substances, especially in the field of the preparation of pharma-

ceuticals and agrochemicals. For example, o,o-dialkylarylketamine derivatives, especially those having alkyl and/or alkoxyalkyl groups, are effective as fungicides, especially as herbicides. The derivatives may be amine salts, acid amides, for example of chloroacetic acid, tertiary amines and ammonium salts (see, for example, EP-A-0 077 755 and EP-A-0 115 470).

Especially important in this connection are the optically active amines of formula



which can be prepared from the imines of formula (V) using the processes according to the invention, wherein  $R_{01}$ ,  $R_{02}$  and  $R_{03}$  are each independently of the others  $C_1$ - $C_4$ alkyl, and  $R_{04}$  is  $C_1$ - $C_4$ alkyl or  $C_1$ - $C_4$ alkoxymethyl or  $C_1$ - $C_4$ alkoxyethyl, and especially the amines of the formulae



which can be prepared from the imines of the formulae (Va) and (Vb) and which can be converted in accordance with methods that are customary *per se* with chloroacetic acid into the desired herbicides of the chloroacetanilide type.

The Examples that follow illustrate the invention in more detail. The chemical conversion is determined by gas chromatography [DB 17/30 W column (15 m), manufacturer: JCW Scientific Inc. USA, temperature programme: from 60°C/1 min to 220°C,  $\Delta T$ : 10° x min<sup>-1</sup>]. The optical yields (enantiomeric excess, ee) are determined either by gas chromatography [Chirasil-Val column, 50 m, manufacturer: Alltech, USA, T = 150°C, isothermic], by HPLC (Chiracel OD column) or by <sup>1</sup>H-NMR spectroscopy (using shift reagents).

Example 1: Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethyl-

amine.

17.2 mg (0.027 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethylphenyl)phosphine and 40 mg (0.108 mmol) of tetrabutylammonium iodide are introduced in succession into a solution of 8.8 mg (0.013 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub> in 10 ml of acetic acid (degassed) and stirred for 15 minutes. Separately, 412 g (2 mol) of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)eth-1-ylideneamine are dissolved in 70 ml of acetic acid (degassed). The imine solution and the catalyst solution are transferred in succession to a 1000 ml steel autoclave which is under an inert gas. In four cycles (10 bar, normal pressure) the inert gas is displaced by hydrogen. Then a pressure of 80 bar of hydrogen is applied and the autoclave heated to 50°C. After a reaction time of 18 hours, the reaction is discontinued and the reaction solution is cooled to room temperature. The hydrogen is depressurised and the reaction solution is expelled under pressure from the autoclave. The conversion is 100 %. 100 ml of toluene are added and then toluene and acetic acid are removed in a rotary evaporator. The residue is distilled under a high vacuum (0.1 mbar), yielding 401 g (yield of 97 %) of pure N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylamine. A sample (2 g) is purified by means of flash chromatography [silica gel 0.040-0.063 mm, eluant hexane/-ethyl acetate 10:1] in order to determine the enantiomeric purity. The optical yield is 75.6 % (S).

Example 2: Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylamine

10.4 mg (0.0155 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub>, 21.4 mg (0.0335 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethylphenyl)phosphine and 50 mg (0.136 mmol) of tetrabutylammonium iodide are dissolved in 2.5 ml of degassed acetic acid and stirred for 15 minutes. Separately, 17 g (0.083 mol) of 2-methyl-6-ethylaniline are dissolved in 9 g of anhydrous methoxyacetone. The methoxyacetone solution and the catalyst solution are transferred in succession to a 50 ml steel autoclave which is under an inert gas. In four cycles (10 bar, normal pressure) the inert gas is displaced by hydrogen. Then a pressure of 40 bar of hydrogen is applied and the autoclave is heated to 50°C. After a reaction time of 18 hours, the reaction is discontinued and the reaction solution is cooled to room temperature. Working up is effected in accordance with Example 1. The conversion is 97 % (based on 2-methyl-6-ethylaniline) and the optical yield is 75.6 % (S).

Example 3: Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethyl-

amine

14.0 mg (0.032 mmol) of (2S,4S)-bis(diphenylphosphino)pentane (BDPP), 70 mg (0.19 mmol) of tetrabutylammonium iodide and 0.3 ml of methanesulfonic acid are introduced in succession into a solution of 10.2 mg (0.015 mmol) of  $[\text{Ir}(1,5\text{-cyclooctadiene})\text{-Cl}]_2$  in 3.5 ml of toluene (degassed) and stirred for 5 minutes. Separately, 3.12 g (15.2 mmol) of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylideneamine are dissolved in 3.2 ml of toluene (degassed). With the aid of a steel capillary the imine solution and the catalyst solution are transferred in succession to a 50 ml steel autoclave which is under an inert gas. In four cycles (10 bar, normal pressure) the inert gas is displaced by hydrogen. Then a pressure of 30 bars of hydrogen is applied. After a reaction period of 2.5 hours at 25°C the reaction is discontinued. Working up is effected in accordance with Example 1. Conversion is 100 % and the optical yield is 53.5 % (R).

**Example 4:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylamine

The process is carried out analogously to Example 3 and the reaction conditions are modified as follows:

0.35 ml of trifluoroacetic acid (instead of methanesulfonic acid). The reaction time is 2 hours, the conversion is 95 % and the optical yield is 52.6 % (R).

**Example 5:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylamine

The process is carried out analogously to Example 3 and the reaction conditions are modified as follows:

0.4 g of *ortho*-phosphoric acid (instead of methanesulfonic acid) and 6.6 ml of tetrahydrofuran as solvent. The reaction time is 2.5 hours, the conversion is 98 % and the optical yield is 53.4 % (R).

**Example 6:** Preparation of N-(2',4'-dimethylthiophen-3'-yl)-N-(1-methoxymethyl)ethylamine

8.6 mg (0.0125 mmol) of  $[\text{Ir}(1,5\text{-cyclooctadiene})\text{Cl}]_2$ , 17.2 mg (0.0268 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethylphenyl)phosphine (ligand) and 30 mg (0.08 mmol) of tetrabutylammonium iodide are introduced in succession into a 10 ml Schlenk flask which is under an argon atmosphere. 1 g (5 mmol) of N-(2',4'-dimethylthiophen-3'-yl)-N-(1-methoxymethyl)ethylideneamine, 5 ml of toluene and 2 ml of acetic acid are added thereto. That solution is transferred by means of

a steel capillary to a 50 ml steel autoclave which is under argon. Then a pressure of 30 bar of hydrogen is applied as described in Example 1 and then the reaction solution is stirred for 20 minutes at room temperature. The reaction is discontinued, the hydrogen is depressurised and the reaction solution is expelled under pressure from the autoclave. Conversion is 100 %. The solvent (toluene) and acid additive (acetic acid) are removed in a rotary evaporator, yielding 1.2 g of oily crude product, which is then purified by flash chromatography (silica gel 0.040 - 0.063 mm, eluant hexane/ethyl acetate (3:1)). The enantiomeric purity of the isolated product is 76.1%.

**Example 7: Preparation of N-(2',4'-dimethylthiophen-3'-yl)-N-(1-methoxymethyl)ethylamine**

The process is carried out as in Example 6, but the reaction conditions are modified as follows:

ligand: 12.3 mg (0.028 mmol) of (2S,4S)-bis(diphenylphosphino)pentane, 55 mg (0.149 mmol) of tetrabutylammonium iodide. The reaction time is 1.4 hours. The conversion is complete, the ee is 47.5%.

**Example 8: Preparation of N-benzyl-N-(1-phenylethyl)amine**

The process is carried out as in Example 6, but the reaction conditions are modified as follows: 0.636 g (3 mmol) of N-benzyl-N-(1-phenylethylidene)amine, 10.2 mg (0.0152 mmol) of  $[\text{Ir}(1,5\text{-cyclooctadiene})\text{Cl}]_2$ , 21.4 mg (0.0333 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethylphenyl)phosphine (ligand) and 15 mg (0.04 mmol) of tetrabutylammonium iodide, 2 ml of acetic acid, 15 ml of toluene, 30 bar of hydrogen, reaction temperature: 25°C. The reaction time is 20 minutes. The conversion is complete, the ee 31%.

**Example 9: Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylamine**

2.7 mg (0.004 mmol) of  $[\text{Ir}(1,5\text{-cyclooctadiene})\text{Cl}]_2$  and 5.8 mg (0.009 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethylphenyl)phosphine are weighed into a Schlenk flask, and then the Schlenk flask is placed under an argon atmosphere. Using a syringe, 2 ml of degassed tetrahydrofuran are then added and the orange solution is stirred for 30 minutes. 210 g (2 mol) of high-purity MEA-imine (>99%) of formula (Vb), 300 mg (0.8 mmol) of tetrabutylammonium iodide and 200 ml of acetic acid are introduced into a 1 litre laboratory autoclave. Then, using a syringe, 0.5 ml of the above catalyst solution is added. The ratio of imine/Ir is 1 000 000. The autoclave is

closed and flushed first with nitrogen, then with hydrogen. Then a pressure of 80 bar of hydrogen is applied and the reaction solution is stirred for 65 hours at a temperature of 50°C internal temperature. When the absorption of hydrogen is complete, the hydrogen is depressurised and then the reaction solution is analysed. The conversion is 100 %, the enantioselectivity 75% ee (S).

**Examples 10 - 22: Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylamine**

In Examples 10 to 22, the process is carried out analogously to Example 6, but with the following modified reaction conditions: 105 g (0.5 mol) of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylideneamine, 1.7 mg (0.0025 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub>, 3.8 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethylphenyl)phosphine, 70 mg (0.189 mmol) of tetrabutylammonium iodide, 80 bar of hydrogen and 50°C. The acids used and the results of the respective tests are shown in Table 1.

Table 1:

Example	acid (g)	time [hrs]	conversion [%]	ee [%]
10	CH <sub>3</sub> COOH (1 g)	16	100	70 (S)
11	Cl <sub>2</sub> CHCOOH (1 g *)	1.5	100	75 (S)
12	Cl <sub>3</sub> CCOOH (1 g *)	1.75	100	76 (S)
13	CH <sub>3</sub> COOH (10 g)	2	100	76 (S)
14	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> COOH (10 g)	3	100	76 (S)
15	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> COOH (10 g)	16	100	76 (S)
16	(CH <sub>3</sub> ) <sub>2</sub> CHCH <sub>2</sub> COOH (10 g)	25	100	76 (S)
17	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> COOH (10 g)	8	95	76 (S)
18	C <sub>6</sub> H <sub>5</sub> COOH (10 g)	19	90	75 (S)
19	CH <sub>3</sub> SO <sub>3</sub> H (1 g)	2.5	100	76 (S)
20	CH <sub>3</sub> P(O)(OH) <sub>2</sub> (1 g)	2	100	76 (S)
21	HOOC(CH <sub>2</sub> ) <sub>2</sub> COOH (10 g)**)	20	90	74 (S)
22	HOOC(CHOH) <sub>2</sub> COOH (10 g)**)	22	91	72 (S)



\*) dissolved in 2 ml of isopropanol

\*\*) suspension in 10 ml of isopropanol

**Example 23:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 4.3 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dipropylaminophenyl)phosphine. The reaction time is 3.5 hours, the conversion: 100%, the enantiomeric purity 83% (S).

**Example 24:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 4.2 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-diisopropyl-4-N,N-dimethylaminophenyl)phosphine. The reaction time is 24 hours, the conversion: 98%, the enantiomeric purity 66% (S).

**Example 25:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 5.0 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-diisopropyl-4-N,N-dibenzylaminophenyl)phosphine. The reaction time is 22 hours, the conversion: 99.5%, the enantiomeric purity 63% (S).

**Example 26:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 4.8 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-N,N-dibenzylaminophenyl)phosphine. The reaction time is 24 hours, the conversion: 85%, the enantiomeric purity 76% (S).

**Example 27:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 4.1 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-4-(1'-pyrrolo)phenyl)phosphine. The reaction time is 3 hours, the conversion:

100%, the enantiomeric purity 69% (S).

**Example 28:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 4.6 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl]}ethyl-di(3,5-dimethyl-4-N,N-dipentylaminophenyl)phosphine. The reaction time is 21 hours, the conversion: 90%, the enantiomeric purity 82.5% (S).

**Example 29:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligands: 4.0 mg (0.0059 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl]}ethyl-di(3,5-dimethyl-4-N,N-dimethylaminophenyl)phosphine. The reaction time is 1 hour, the conversion: 100%, the enantiomeric purity 80% (S).

**Example 30:** Preparation of N-benzyl-N-(1-phenylethyl)amine

The process is carried out as in Example 8 but the reaction conditions are modified as follows: 0.636 g (4.8 mmol) of N-benzyl-N-(1-phenylethylidene)amine, 3.2 mg (0.0048 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub>, 4.5 mg (0.01 mmol) of 1,4-bis(diphenylphosphino)butane (ligand) and 30 mg (0.08 mmol) of tetrabutylammonium iodide, 2 ml of acetic acid, 5 ml of toluene, 40 bar of hydrogen, reaction temperature: 25°C. The reaction time is 2 hours, the conversion is complete.

**Example 31:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following ligand: 4.1 mg (0.0059 mmol) of {(R)-1-[(S)-2-di(4-methoxyphenyl)phosphino]ferrocenyl]}ethyl-di(3,5-dimethyl-4-N,N-dimethylaminophenyl)phosphine. The reaction time is 3.5 hours, the conversion: 100%, the enantiomeric purity 76% (S).

**Example 32:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following catalyst precursors instead of the *in situ* catalyst: 10.4 mg (0.01 mmol) of [Ir(1,5-cyclooctadiene)-{(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}]ethyl-di(3,5-dimethylphenyl)phosphine]-

BF<sub>4</sub>, 135 mg (0.365 mmol) of tetrabutylammonium iodide; 0.3 litre steel autoclave. The reaction time is 45 min, the conversion: 100%, the enantiomeric purity 78% (S).

**Example 33:** Preparation of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)-ethylamine

The process is carried out analogously to Example 13, but using the following catalyst precursors instead of the *in situ* catalyst: 9.9 mg (0.01 mmol) of [Ir(1,5-cyclooctadiene)-((R)-1-[(S)-2-diphenylphosphino]ferrocenyl)]ethyl-di(3,5-dimethylphenyl)phosphine)-Cl], 135 mg (0.365 mmol) of tetrabutylammonium iodide; 0.3 litre steel autoclave. The reaction time is 35 min, the conversion: 100%, the enantiomeric purity 77.8% (S).

**Example 34:** Preparation of N-(2',6'-dimethylphen-1'-yl)-N-(1-methoxymethyl)ethylamine

The process is carried out as in Example 6 but the reaction conditions are modified as follows: 514 g (2.6 mol) of N-(2',6'-dimethylphen-1'-yl)-N-(1-methoxymethyl)ethylamine, 77 mg (0.115 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub>, 214 mg (0.27 mmol) of {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl]}ethyl-di(3,5-dimethyl-4-N,N-dipropylamino-phenyl)phosphine, 3.5 g (9.5 mmol) of tetrabutylammonium iodide, 50 ml of acetic acid, 80 bar of hydrogen, temperature: 50-60°C. The reaction time is 2.5 hours, the conversion: 100 %, the enantiomeric purity 78.9 % (S).

**Example 35:** Preparation of N-(2',6'-dimethylphen-1'-yl)-N-(1-methoxymethyl)ethylamine

The process is carried out as in Example 33 but the reaction conditions are modified as follows:

5 ml (0.024 mol) of N-(2',6'-dimethylphen-1'-yl)-N-(1-methoxymethyl)ethylamine, 10.2 mg (0.015 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub>, 21.5 mg (0.033 mmol) of {(R)-1[(S)-2-diphenylphosphino]ferrocenyl]}ethyl-di(3,5-dimethylphenyl)phosphine, 50 mg (0.135 mmol) of tetrabutylammonium iodide, 2 ml of acetic acid, 80 bar of hydrogen, temperature: 50-60°C, 50 ml small autoclave. The reaction time is 1 hour, the conversion: 100 %, the enantiomeric purity 56.2 % (S).

**Example 36:**

The procedure followed is analogous to Example 6 but with the following modified reaction conditions.

31 kg (148.3 mol) of N-(2'-methyl-6'-ethyl-phen-1'-yl)-N-(1-methoxymethyl)ethylidene-amine are placed in a 50 litre steel autoclave, followed by the addition of 500 mg (0.744 mmol) of [Ir(1,5-cyclooctadiene)Cl]<sub>2</sub>, 1.15 g (1.8 mmol) of {(R)-1-[(S)-2-diphenyl-

phosphino)ferrocenyl]]ethyl-di(3,5-dimethylphenyl)phosphine, 22.5 g (61 mmol) of tetrabutylammonium iodide and 3 litres of acetic acid. The hydrogen pressure is 75 bar, the reaction temperature 50°C. After a reaction time of 13 hours the conversion is complete. The ee is 75 % (S).

**Example 37: Preparation of S-2-chloro-N-(2,6-dimethylphenyl)-N-(2-methoxy-1-methylethyl)-acetamide**

With stirring and while passing nitrogen through the mixture, 433 g (5.48 mol) of pyridine are added dropwise at 15-20°C in the course of 25 minutes to a solution of 883 g (4.57 mol) of S-2,6-dimethyl-N-(2-methoxy-1-methylethyl)-aniline (ee 78.2 %) in 1.8 litres of toluene. Then, with ice-cooling at 15-20°C, 547 g (4.84 mol) of chloroacetyl chloride are added dropwise thereto in the course of 1.5 hours. When the dropwise addition is complete, the suspension so obtained is stirred for a further 1.5 hours at room temperature. For working-up, the reaction mixture is poured onto 2 litres of water and extracted twice using 200 ml of toluene each time. The organic phases are combined, washed once with 300 ml of 2N hydrochloric acid, twice using 300 ml of saturated sodium chloride solution each time and once with 600 ml of saturated sodium hydrogen carbonate solution, dried over sodium sulfate and filtered, and the solvent is removed *in vacuo*. For purification, the crude product so obtained is subjected to fractional distillation. B.p.<sub>0.3</sub> 138-140°C; ee 78.1 %.

**Example 38: Preparation of S-2-chloro-N-(2-ethyl-6-methylphenyl)-N-(2-methoxy-1-methylethyl)-acetamide**

10.52 kg (50.7 mol) of S-2-ethyl-N-(2-methoxy-1-methylethyl)-6-methylaniline (ee 80.9 %;  $[\alpha]_D^{20}$ : 16.43 c: 2.6112 in hexane) are placed in 20 litres of toluene, and at 10°C 4812 g (60.8 mol) of pyridine are added. With ice-cooling at 10-20°C, 6073 g (53.7 mol) of chloroacetyl chloride are then added dropwise in the course of 2.5 hours to the reaction solution so obtained. When the addition is complete, the resulting suspension is stirred at room temperature for 16 hours. For working-up, the reaction mixture is poured onto 20 litres of water and the resulting emulsion is stirred vigorously for 10 minutes. After removal of the organic phase, the aqueous phase is extracted once with 10 litres of hexane. The combined organic phases are washed once with 10 litres of water, once with 5 litres of 2N hydrochloric acid and once with 10 litres of water, dried over sodium sulfate and filtered, and concentrated in a rotary evaporator. For purification, the crude product so obtained is subjected to fractional distillation. B.p. <sub>0.1</sub> 135-140°C; ee 81.0 %;  $[\alpha]_D^{20}$ : -6.53 c: 2.2364 in hexane.

Example 39:

40 kg (194 mol) of MEA-imine of formula (Vb) are drawn into an inert container towards a closed vacuum and the residual vacuum is broken with nitrogen. The contents of the container are then introduced under pressure into an inert 50 litre loop reactor (loop reactor manufactured by Buss). After a mixture of

0.14 g ( $2.08 \cdot 10^{-4}$  mol) of  $[\text{Ir}(\text{COD})\text{Cl}]_2$

0.27 g ( $4.23 \cdot 10^{-4}$  mol) of ligand {(R)-1-[(S)-2-diphenylphosphino]ferrocenyl}ethyl-di(3,5-dimethyl-phenyl)phosphine and

6.20 g ( $1.66 \cdot 10^{-2}$  mol) of TBAI (tetrabutylammonium iodide),

has been introduced into the reactor *via* a solids sluice, rinsing is carried out with 4.1 kg (68 mol) of acetic acid (anhydrous) and the reactor is depressurised. The reactor is then twice pressurised with hydrogen to 5 bar and depressurised. The accompanying heating of the reactor is set to  $T_a = 50^\circ\text{C}$ . The loop reactor is then pressurised to 80 bar with hydrogen and the circulating pump is switched on. A rapid absorption of hydrogen is observed which achieves the theoretical hydrogen consumption after about 1-2 hours. When no further absorption of hydrogen can be detected, the reactor contents are cooled to room temperature and depressurised. The reactor is then rendered inert with nitrogen and the contents are removed. The hydrogenated solution is worked up by distillation and the product is isolated in a yield of 98 %.

Example 40:

The process is carried out as in Example 6, but the reaction conditions are modified as follows: in a 1l-reaction vessel 413 g (2,004 mmol) of the imine, 2,8 mg of the iridium compound, 6,4 mg of the diphosphine ligand and 124,2 mg of the iodide are used. Instead of 2 ml of acetic acid 0,1 g of  $\text{H}_2\text{SO}_4$  are used.

The conversion is 100 %. After isolation and purification according to Example 6 one obtains 99 % of the desired product, the optical yield being 76,0 %(S).